9/13/16



CARBON DIOXIDE Safety Data Sheet

Amerex Co2 Fire Extinguisher

1. IDENTIFICATION

Product identifier

Product Name

CARBON DIOXIDE

Other means of identification

Safety data sheet number

LIND-P023

UN/ID no.

UN1013

Synonyms

LASER Carbon Dioxide, LASER Carbon Dioxide Ultra, Carbonic Anhydride, Carbonic Acid Gas, Carbon

Dioxide USP

Recommended use of the chemical and restrictions on use

Recommended Use

Industrial and professional use.

Uses advised against

Consumer use

Details of the supplier of the safety data sheet

Linde Gas North America LLC - Linde Merchant Production Inc. - Linde LLC

575 Mountain Ave. Murray Hill, NJ 07974 Phone: 908-464-8100 www.lindeus.com

Linde Gas Puerto Rico, Inc.

Road 869, Km 1.8

Barrio Palmas, Catano, PR 00962

Phone: 787-641-7445 www.pr.lindegas.com

Linde Canada Limited 5860 Chedworth Way Mississauga, Ontario L5R 0A2 Phone: 905-501-1700 www.lindecanada.com

For additional product information contact your local customer service.

Emergency telephone number

Company Phone Number

800-232-4726 (Linde National Operations Center, US)

905-501-0802 (Canada)

CHEMTREC: 1-800-424-9300 (North America) +1-703-527-3887 (International)

^{*} May include subsidiaries or affiliate companies/divisions.

2. HAZARDS IDENTIFICATION

Classification

OSHA Regulatory Status

This chemical is considered hazardous by the 2012 OSHA Hazard Communication Standard (29 CFR 1910.1200).

Gases under pressure		Liquefied gas	
Simple asphyxiants		Yes	

Label elements



Signal word

Warning

Hazard Statements Contains gas under pressure; may explode if heated May displace oxygen and cause rapid suffocation

May cause frostbite May increase respiration and heart rate

Precautionary Statements - Prevention
Do not handle until all safety precautions have been read and understood
Avoid breathing gas
Do not get in eyes, on skin, or on clothing
Use and store only outdoors or in a well ventilated place
Use a backflow preventive device in piping
Use only with equipment rated for cylinder pressure
Close valve after each use and when empty

Precautionary Statements - Response

IF INHALED: Remove person to fresh air and keep comfortable for breathing. Get medical attention/advice. IF ON SKIN:. Get immediate medical advice/attention. Thaw frosted parts with lukewarm water. Do not rub affected area.

Precautionary Statements - Storage
Protect from sunlight when ambient temperature exceeds 52°C/125°F

<u>Hazards not otherwise classified (HNOC)</u> Not applicable

3. COMPOSITION/INFORMATION ON INGREDIENTS

			and the second s
Chemical Name	CAS No.	Volume %	Chemical Formula
Carbon dioxide	124-38-9	100	CO ₂

4. FIRST AID MEASURES

Description of first aid measures

General advice Show this safety data sheet to the doctor in attendance.

Inhalation Remove to fresh air and keep comfortable for breathing. If breathing is difficult, give oxygen. If

breathing has stopped, give artificial respiration. Get medical attention immediately.

Skin contact For dermal contact or suspected frostbite, remove contaminated clothing and flush affected areas

with lukewarm water. DO NOT USE HOT WATER. A physican should see the patient promptly if contact with the product has resulted in blistering of the dermal surface or in deep tissue freezing.

Eye contact If frostbite is suspected, flush eyes with cool water for 15 minutes and obtain immediate medical

attention.

Ingestion Not an expected route of exposure.

Self-protection of the first aider RESCUE PERSONNEL SHOULD BE EQUIPPED WITH SELF-CONTAINED BREATHING APPARATUS.

Most important symptoms and effects, both acute and delayed

Symptoms Simple asphyxiant. May cause suffocation by displacing the oxygen in the air. Exposure to

oxygen-deficient atmosphere (<19.5%) may cause dizziness, drowsiness, nausea, vomiting, excess salivation, diminished mental alertness, loss of consciousness and death. Exposure to atmospheres containing 8-10% or less oxygen will bring about unconsciousness without warning and so quickly that the individuals cannot help or protect themselves. Lack of sufficient oxygen may cause serious injury or death. Depending on concentration and duration of exposure to carbon dioxide may cause increased respirations, headache, mild narcotic effects, increased blood pressure and pulse, and asphyxiation. Symptoms of overexposure become more apparent when atmospheric oxygen is

decreased to 15-17%. Contact with liquid may cause cold burns/frostbite.

Indication of any immediate medical attention and special treatment needed

5. FIRE-FIGHTING MEASURES

Suitable extinguishing media

Use extinguishing measures that are appropriate to local circumstances and the surrounding environment.

Specific extinguishing methods

Continue to cool fire exposed cylinders until flames are extinguished. Damaged cylinders should be handled only by specialists.

Specific hazards arising from the chemical

Non-flammable gas. Cylinders may rupture under extreme heat.

Protective equipment and precautions for firefighters

As in any fire, wear self-contained breathing apparatus pressure-demand, MSHA/NIOSH (approved or equivalent) and full protective gear.

6. ACCIDENTAL RELEASE MEASURES

Personal precautions, protective equipment and emergency procedures

Personal precautions Evacuate personnel to safe areas. Ensure adequate ventilation, especially in confined areas.

Monitor oxygen level. Wear self-contained breathing apparatus when entering area unless

atmosphere is proved to be safe.

Environmental precautions

Environmental precautions

Prevent spreading of vapors through sewers, ventilation systems and confined areas.

Methods and material for containment and cleaning up

Methods for containment

Stop the flow of gas or remove cylinder to outdoor location if this can be done without risk. If leak is in container or container valve, contact the appropriate emergency telephone number in Section 1 or call your closest Linde location.

Methods for cleaning up

Return cylinder to Linde or an authorized distributor.

7. HANDLING AND STORAGE

Precautions for safe handling

Advice on safe handling

For applications with moist Carbon Dioxide, 316, 309 and 310 stainless steels may be used as well as Hastelloy® A, B, & C and Monel®. Ferrous nickel alloys are slightly suspectible to corrosion. At normal temperatures carbon dioxide is compatible with most plastics and elastomers.

Protect cylinders from physical damage; do not drag, roll, slide or drop. When moving cylinders, even for short distance, use a cart designed to transport cylinders. Never attempt to lift a cylinder by its valve protection cap. Never insert an object (e.g. wrench, screwdriver, pry bar,etc.) into valve cap openings. Doing so may damage valve, causing leak to occur. Use an adjustable strap wrench to remove over-tight or rusted caps. Use only with adequate ventilation. Use a backflow preventive device in piping. Use only with equipment rated for cylinder pressure. Close valve after each use and when empty. If user experiences any difficulty operating cylinder valve discontinue use and contact supplier. Ensure the complete gas system has been checked for leaks before use.

Never put cylinders into trunks of cars or unventilated areas of passenger vehicles. Never attempt to refill a compressed gas cylinder without the owner's written consent. Never strike an arc on a compressed gas cylinder or make a cylinder a part of an electrical circuit.

Only experienced and properly instructed persons should handle gases under pressure. Always store and handle compressed gas cylinders in accordance with Compressed Gas Association, pamphlet CGA-P1, Safe Handling of Compressed Gases in Containers.

For additional storage recommendations, consult Compressed Gas Association's Pamphlets P-1,AV-7, G-6, G-6.1, G-6.2, G6.3, G-6.5, G-6.7, G-6.9, PS-5,TB-10, and SB-2.

Conditions for safe storage, including any incompatibilities

Storage Conditions

Store in cool, dry, well-ventilated area of non-combustible construction away from heavily trafficked areas and emergency exits. Keep at temperatures below 52°C / 125°F. Cylinders should be stored upright with valve protection cap in place and firmly secured to prevent falling. Full and empty cylinders should be segregrated. Use a "first in-first out" inventory system to prevent full cylinders from being stored for excessive periods of time. Stored containers should be periodically checked for general condition and leakage.

Incompatible materials

Certain reactive metals, hydrides, moist cesium monoxide, or lithium acetylene carbide diammino may ignite. Passing carbon dioxide over a mixture of sodium peroxide and aluminum or magnesium may explode.

8. EXPOSURE CONTROLS/PERSONAL PROTECTION

Control parameters

Exposure Guidelines

Chemical Name	ACGIH TLV	OSHA PEL	NIOSH IDLH
Carbon dioxide	STEL = 30000 ppm	TWA: 5000 ppm	IDLH: 40000 ppm
124-38-9	TWA: 5000 ppm	TWA: 9000 mg/m ³	TWA: 5000 ppm
		(vacated) TWA: 10000 ppm	TWA: 9000 mg/m ³
		(vacated) TWA: 18000 mg/m³	STEL: 30000 ppm
		(vacated) STEL: 30000 ppm	STEL: 54000 mg/m ³
		(vacated) STEL: 54000 mg/m³	

ACGIH TLV: American Conference of Governmental Industrial Hygienists - Threshold Limit Value. OSHA PEL: Occupational Safety and Health Administration - Permissible Exposure Limits. NIOSH IDLH: Immediately Dangerous to Life or Health

Other Information

Vacated limits revoked by the Court of Appeals decision in AFL-CIO v. OSHA, 965 F.2d 962 (11th Cir.,

1992).

Appropriate engineering controls

Engineering Controls

Local exhaust ventilation to prevent accumulation of high concentrations and maintain air-oxygen levels at or above 19.5%. Oxygen detectors should be used when asphyxiating gases may be released. Systems under pressure should be regularly checked for leakages.

Individual protection measures, such as personal protective equipment

Eye/face protection Wear safety glasses with side shields (or goggles). If splashes are likely to occur, wear:. Goggles.

Face-shield.

Skin and body protection Work gloves and safety shoes are recommended when handling cylinders. Wear cold insulating

gloves when handling liquid.

Respiratory protection Use positive pressure airline respirator with escape cylinder or self contained breathing apparatus

for oxygen-deficient atmospheres (<19.5%). If exposure limits are exceeded or irritation is experienced, NIOSH/MSHA approved respiratory protection should be worn. Positive-pressure supplied air respirators may be required for high airborne contaminant concentrations. Respiratory

protection must be provided in accordance with current local regulations.

General Hygiene Considerations Handle in accordance with good industrial hygiene and safety practice. Do not get in eyes, on skin,

or on clothing.

9. PHYSICAL AND CHEMICAL PROPERTIES

Information on basic physical and chemical properties

Physical state

Appearance Odor

Odor threshold

pН

Melting point Evaporation rate

Lower flammability limit: Upper flammability limit:

Flash point

Autoignition temperature Decomposition temperature

Water solubility
Partition coefficient
Kinematic viscosity

Compressed liquefied gas

Colorless.

No information available No data available

-56.6 °C / -69.8 °F Not applicable Not applicable

Not applicable Not applicable No data available No data available

0.145 g/ml @ 25°C No data available Not applicable

Component Level Information:

	component Ecver Information.									
-	Chemical Name	Molecular weight	Boiling point	Vapor Pressure	Vapor density (air	Gas Density	Critical			
١					=1)	kg/m³@20°C	Temperature			

				· · · · · · · · · · · · · · · · · · ·		1.
Carbon dioxide	44.01	-78.5 °C	838 psig (5778	1.522	1.839	31.1 °C
		(Sublimes)	kPa) @ 21.1°C			

10. STABILITY AND REACTIVITY

Not reactive under normal conditions

Chemical stability

Stable under normal conditions.

Explosion data

Sensitivity to Mechanical Impact

None.

Sensitivity to Static Discharge

None.

Possibility of Hazardous Reactions

None under normal processing.

Conditions to avoid

Due to the presence of Carbon dioxide, Carbonic acid is formed in the presence of moisture.

Incompatible materials

Certain reactive metals, hydrides, moist cesium monoxide, or lithium acetylene carbide diammino may ignite. Passing carbon dioxide over a mixture of sodium peroxide and aluminum or magnesium may explode.

Hazardous Decomposition Products

Oxygen. Carbon monoxide (CO).

11. TOXICOLOGICAL INFORMATION

Information on likely routes of exposure

Inhalation Acidosis, adrenal cortical exhaustion, and other metabolic stresses have resulted from prolonged

> continuous exposure to 1-2% carbon dioxide (10,000 ppm-20,000 ppm). The ACGIH TLV of 5,000 ppm is expected to provide a good margin of safety from asphyxiation and undue metabolic stress provided sufficient oxygen levels are maintained in the air. Increased physical activity, duration of exposure, and decreased oxygen content can affect systemic and respiratory effects resulting from

exposure to carbon dioxide.

Skin contact Contact with liquid may cause cold burns/frostbite.

Eye contact Contact with liquid may cause cold burns/frostbite.

Ingestion Not an expected route of exposure.

Information on toxicological effects

Symptoms Depending on concentration and duration of exposure to carbon dioxide may cause increased

respirations, headache, mild narcotic effects, increased blood pressure and pulse, and asphyxiation. Symptoms of overexposure become more apparent when atmospheric oxygen is

decreased to 15-17%.

Delayed and immediate effects as well as chronic effects from short and long-term exposure

Irritation Not classified.

Sensitization Not classified. Germ cell mutagenicity Not classified.

Carcinogenicity This product does not contain any carcinogens or potential carcinogens listed by OSHA, IARC or NTP.

Reproductive toxicity Not classified. STOT - single exposure

Not classified.

STOT - repeated exposure

Not classified.

Chronic toxicity

 $Chronic\ harmful\ effects\ are\ not\ known\ from\ repeated\ inhalation\ of\ concentrations\ below\ PEL/TLV.$

Target Organ Effects

Central Vascular System (CVS), Respiratory system.

Aspiration hazard

Not applicable.

Numerical measures of toxicity

Component Level Information:

Chemical Name	Oral LD50	Dermal LD50	Dermal LD50 Inhalation LC50	
Carbon dioxide 124-38-9			470,000 ppm (Rat)	· •

Product Information

Oral LD50 Dermal LD50 Inhalation LC50 No information available

No information available

TCLo - 10,000 ppm (Rat) 24 hours/30 days-continuous

12. ECOLOGICAL INFORMATION

Ecotoxicity

No known acute aquatic toxicity.

Persistence and degradability

No information available.

Bioaccumulation

No information available.

Global warming potential (GWP)

1

13. DISPOSAL CONSIDERATIONS

Waste treatment methods

Disposal of wastes

Do not attempt to dispose of residual waste or unused quantities. Return in the shipping container PROPERLY LABELED WITH ANY VALVE OUTLET PLUGS OR CAPS SECURED AND VALVE PROTECTION CAP

IN PLACE to Linde for proper disposal.

14. TRANSPORT INFORMATION

DOT

UN/ID no.

UN1013

Proper shipping name

Carbon Dioxide

Hazard Class

2.2

Description

UN1013, Carbon dioxide, 2.2

Emergency Response Guide Number 120

<u>TDG</u>

UN/ID no.

UN1013

Proper shipping name

Carbon Dioxide

Hazard Class

2.2

Description UN1013, Carbon dioxide, 2.2

MEX

UN/ID no.

UN1013

Proper shipping name

Carbon Dioxide

Hazard Class

2.2

Description

UN1013, Carbon dioxide, 2.2

IATA

UN/ID no.

UN1013

Proper shipping name

Carbon Dioxide

Hazard Class

2.2 2L

ERG Code Description

UN1013, Carbon dioxide, 2.2

IMDG

UN/ID no.

UN1013

Proper shipping name

Carbon dioxide

Hazard Class

2.2

EmS-No.
Description

F-C, S-V UN1013, Carbon dioxide, 2.2

ADR

UN/ID no.

UN1013

Proper shipping name

Carbon Dioxide

Hazard Class

2.2

Classification code Tunnel restriction code 2A (C/E)

Special Provisions

584, 653

Description

UN1013, Carbon dioxide, 2.2, (C/E)

15. REGULATORY INFORMATION

International Inventories

TSCA

Complies

DSL/NDSL

EINECS/ELINCS

Complies Complies

Legend:

TSCA - United States Toxic Substances Control Act Section 8(b) Inventory

DSL/NDSL - Canadian Domestic Substances List/Non-Domestic Substances List

EINECS/ELINCS - European Inventory of Existing Chemical Substances/European List of Notified Chemical Substances

US Federal Regulations

SARA 313

Section 313 of Title III of the Superfund Amendments and Reauthorization Act of 1986 (SARA). This product does not contain any chemicals which are subject to the reporting requirements of the Act and Title 40 of the Code of Federal Regulations, Part 372.

SARA 311/312 Hazard Categories

Acute Health Hazard No
Chronic Health Hazard No
Fire Hazard No
Sudden release of pressure hazard Yes
Reactive Hazard No

CERCLA

This material, as supplied, does not contain any substances regulated as hazardous substances under the Comprehensive Environmental Response Compensation and Liability Act (CERCLA) (40 CFR 302) or the Superfund Amendments and Reauthorization Act (SARA) (40 CFR 355). There may be specific reporting requirements at the local, regional, or state level pertaining to releases of this material.

Clean Air Act, Section 112 Hazardous Air Pollutants (HAPs) (see 40 CFR 61)

This product does not contain any substances regulated as hazardous air pollutants (HAPS) under Section 112 of the Clean Air Act Amendments of 1990.

CWA (Clean Water Act)

This product does not contain any substances regulated as pollutants pursuant to the Clean Water Act (40 CFR 122.21 and 40 CFR 122.42)

Risk and Process Safety Management Programs

This material, as supplied, does not contain any regulated substances with specified thresholds under 40 CFR Part 68. This product does not contain any substances regulated as Highly Hazardous Chemicals pursuant to the 29 CFR Part 1910.110.

US State Regulations

California Proposition 65

This product does not contain any Proposition 65 chemicals

U.S. State Right-to-Know Regulations

Chemical Name		New Jersey	Massachusetts	Pennsylvania
Carbon dioxide		X	X	Χ.
124-38-9	.1.			

International Regulations

	Chemical Name		Carcinogenicity		Exposure Limits
	Carbon dioxide	1.1			Mexico: TWA= 5000 ppm
and the state of t					Mexico: TWA= 9000 mg/m ³
1					Mexico: STEL= 15000 ppm
				٠.	Mexico: STEL= 27000 mg/m ³

16, OTHER INFORMATION

NFPA

Health hazards 2

Flammability 0

Instability 0

Physical and Chemical Properties Simple asphyxiant

Note: Ratings were assigned in accordance with Compressed Gas Association (CGA) guidelines as published in CGA Pamphlet P-19-2009, CGA Recommended Hazard Ratings for Compressed Gases, 3rd Edition.

Issue Date

17-Feb-2015

Revision Date

17-Apr-2015

Revision Note

Initial Release.

General Disclaimer

For terms and conditions, including limitation of liability, please refer to the purchase agreement in effect between Linde LLC, Linde Merchant Production, Inc. or Linde Gas North America LLC (or any of their affiliates and subsidiaries) and the purchaser.

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End of Safety Data Sheet